

Supporting your child to revise using past papers

Past papers are one of the most effective ways to revise and improve grades.

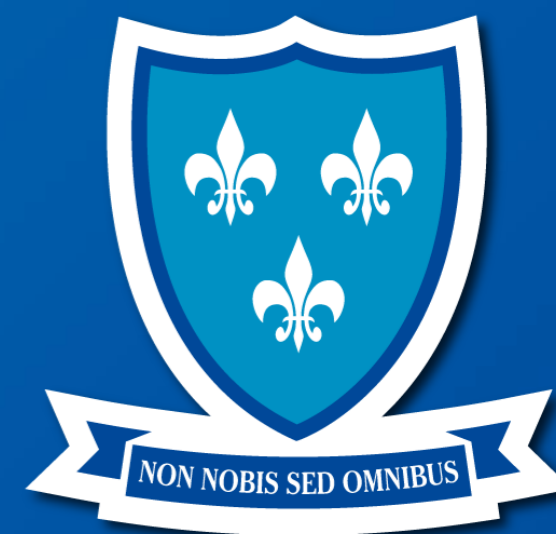
We can use two approaches:

Completing a full question paper

Completing part of a question paper

Full question papers can be accessed at the following address:

<http://www.aqa.org.uk/subjects/science/gcse/combined-science-trilogy-8464/assessment-resources>



When completing past paper questions students are honing their skills so they need to **focus on:**

- their use of **correct terminology**
- that they **match the number of points they make to the marks available**
- that they have the ability to **apply their knowledge** in a variety of **different context.**
- that they make sure they answer **each part of the question.**



Using the markscheme:

When given a list of acceptable answers where more than one mark is available you will see ‘**any two from**’ for example. Each bullet point is a potential mark, *watch out for answers that are from the **same marking point** i.e. bullet point*

0 5 . 7 Suggest **two** reasons why the survival rates for all cancers have increased.

[2 marks]

1 Improved treatments for patients

2 Improved drugs used

05.7	any two from: <ul style="list-style-type: none">• improved treatment / drugs• earlier diagnosis• more cancer screening• improved patient knowledge (of risk factors)	allow improved patient diet / lifestyle	2	AO3/2a 4.2.2.6 4.2.2.7
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A bold **and** is used to indicate that both parts of the answer are required to award the mark and **or** is used to indicate alternative answers, *don't forget to look in the column additional information for more guidance.*

0 6 . 4 Infected mosquitoes landed on the socks three times more often than uninfected mosquitoes.

Explain how this information can be used to reduce the spread of malaria.

[2 marks]

Wear clean socks to stop mosquitos from being attracted

06.4	use worn socks or use chemical from worn socks	or accept: wear clean socks / change socks regularly (1) to reduce the chance of attracting mosquitoes (1)	1	AO3/3a 4.3.1.5
	to attract / trap infected mosquitoes		1	AO3/3a 4.3.1.5



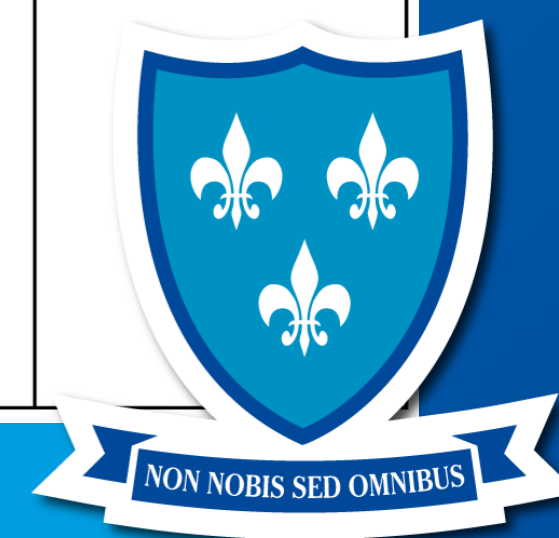
Brackets (...) are used to indicate which bits of information are **not essential** for the mark *but add clarity to the markscheme*.

0 **1** . **6** Describe how an indicator can be used to show when all the sodium hydroxide has reacted with sulfuric acid.

[3 marks]

Put a pH probe into the acid then add the sodium hydroxide
 until it reads 7.

01.6	Add indicator to sodium hydroxide solution	allow add indicator to sulfuric acid	1	AO2/2 5.4.2.2, 4
	Add sulfuric acid (gradually)	allow add sodium hydroxide solution (gradually)	1	
	until indicator just changes (colour) or until universal indicator turns green or shows pH7	allow pH probe	1	



Any wording that is underlined is essential for the marking point to be awarded.

Allow and **accept** means that the response is creditworthy

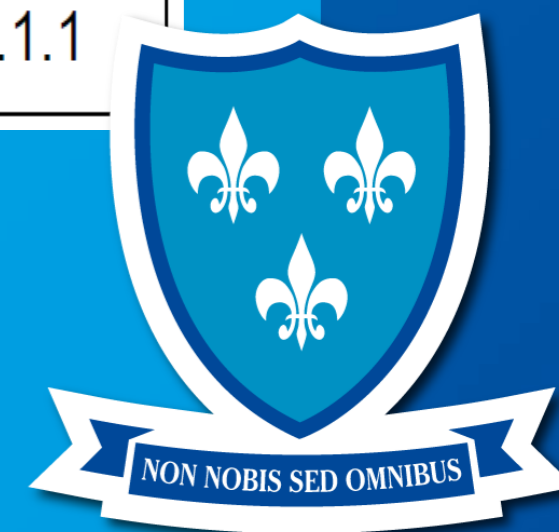
03.2 . 2 A helium atom is much larger than an alpha particle.

Give **one** other difference between a helium atom and an alpha particle.

[1 mark]

It is lighter

03.2	(a helium atom) has 2 <u>electrons</u>	accept it has more mass allow it is not charged	1	AO2/1 6.4.1.1
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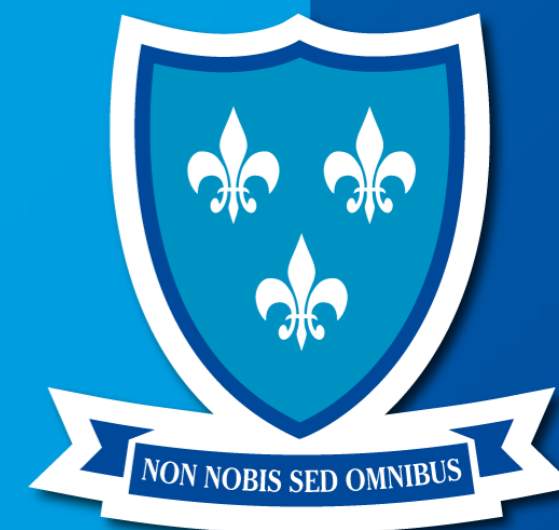


Questions involving chemical formulae, the **case matters** as does the **superscript** and **subscript**.

E.g what is the correct formulae for carbon dioxide?

[1 marks]

<u>Student</u>	<u>Response</u>	<u>Marks awarded</u>
1.	CO2	0
2.	CO ₂	1
3.	Co ₂	0



Questions involving calculations.

Marks are awarded for **each stage of the calculation** so the students are encouraged to **show their working!**

Error carried forward (**e.c.f**) is used when marking questions involving several parts, where the answer to part 1 is used in the answer to part 2.

See handout [1]



Often a **mark** is allocated to writing the answer to the correct **level of precision**.

(i) The teacher used 1.00 g of magnesium.

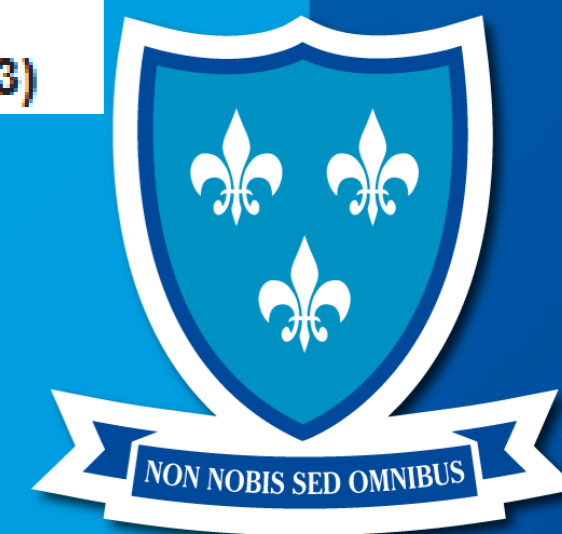
Use the equation to calculate the maximum mass of magnesium oxide produced.

Give your answer to three significant figures.

Relative atomic masses (A_r): O = 16; Mg = 24

Maximum mass = _____ g

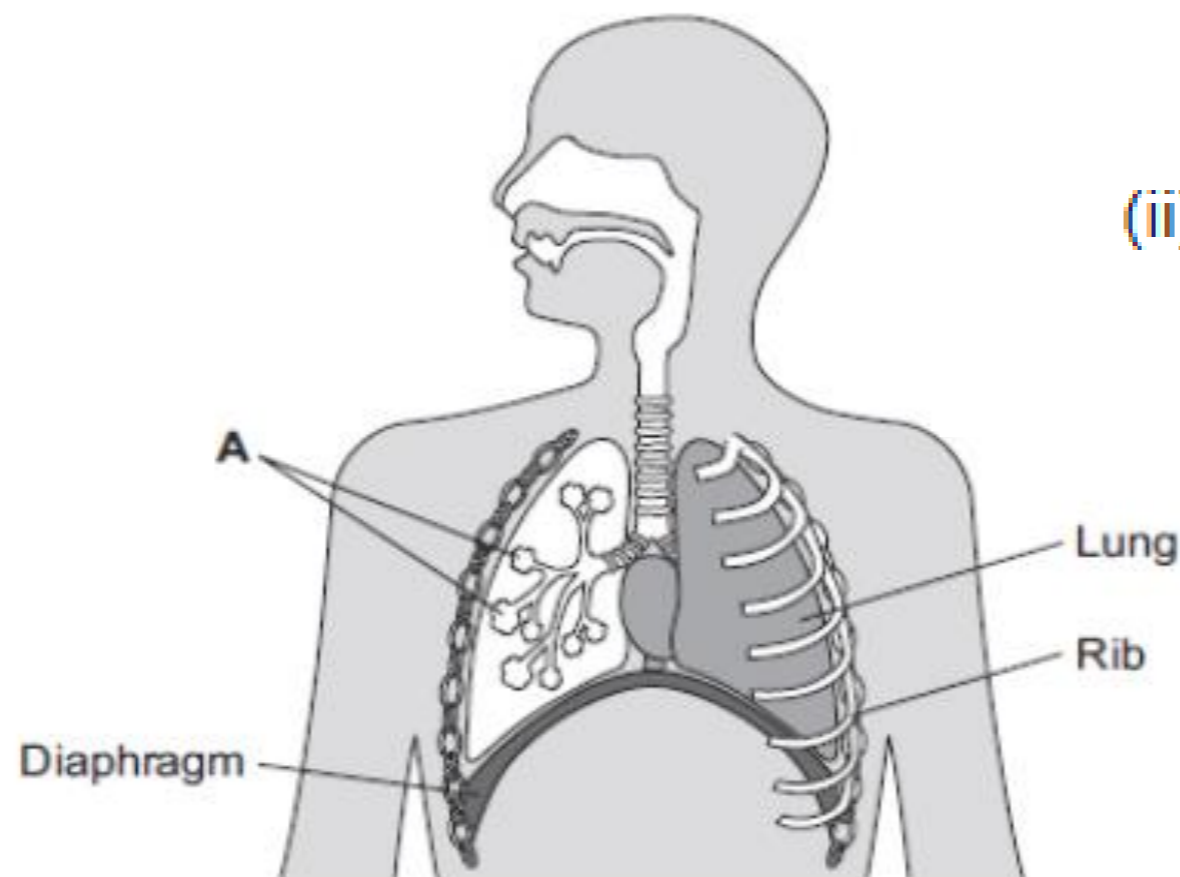
(3)



Phonetic spelling of correct scientific terminology is acceptable unless there is a possible confusion with another technical term.

Our lungs help us to breathe.

The image below shows the human breathing system.



(a) (i) alveoli / alveolus
allow air sacs
allow phonetic spelling

(ii) any **one** from:
• protection (of lungs / heart)
• help you breathe / inflate lungs.

(a) (i) Name part A.

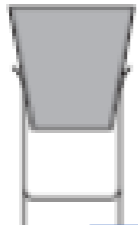
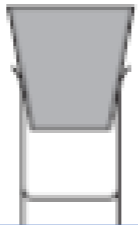
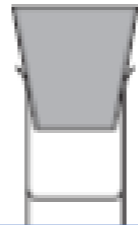
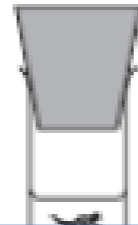
(1)

(ii) Give **one** function of the ribs.

(1)

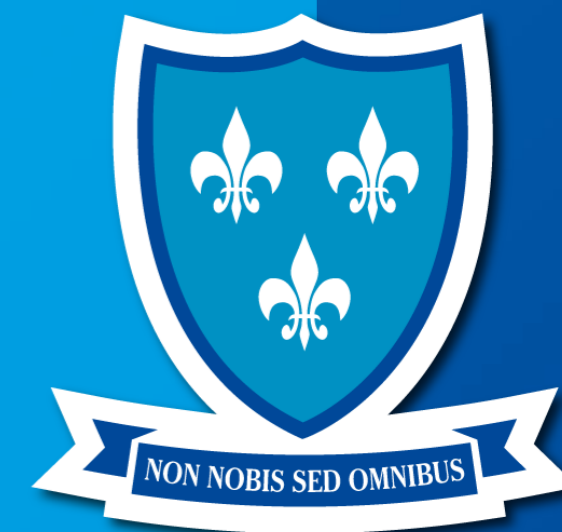


Do not accept means an answer is **wrong** even if the correct answer is also given but **ignore** means that further amplification can still gain full credit.

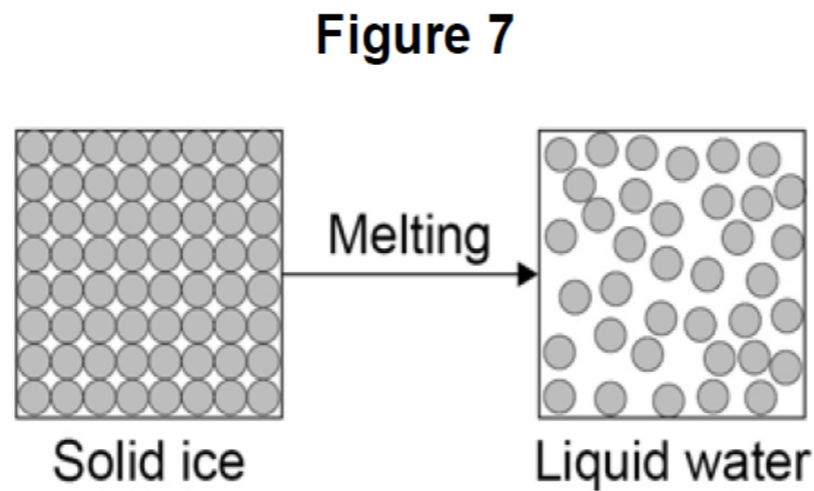
Tube A	Tube B	Tube C	Tube D
			
<p>(b) (i) control</p> <p>do not accept 'control variable'</p> <p>allow:</p> <p>to show the effect of the organisms</p> <p>or</p> <p>to allow comparison</p> <p>or</p> <p>to show the indicator doesn't change on its own</p>			
Indicator only			ion
Stays green	Turns blue	Turns yellow	+ small

(i) What is the purpose of **Tube A**?

(1)



0 5 . 1 Use the particle model to describe how the heating element causes the arrangement of particle in ice to change as the ice melts.



Your should include a description of how the particles are arranged in the solid ice and in the water

[6 marks]

- When tackling this type of question students are encouraged to:
- plan their answer before attempting it, *so that it is presented in a logical order*
 - make sure that they **answer all** parts of the question, *so highlight each part.*

They can bullet point their answers, but they must be in a logical order to gain full credit.



Extended response questions are marked using a **level of response** mark schemes.

05.1	Level 3: A clear, logical explanation containing accurate ideas presented in the correct order with links between ideas.	5–6	6	2 AO2/1
	Level 2: Key ideas presented with some linked together to form a partial explanation.	3–4		2 AO2/1
	Level 1: Fragmented ideas, some may be relevant, insufficient links to form an explanation.	1–2		2 AO1/1 6.3.1.1
	No relevant content.	0		6.2.4.2



Step 1: Determine a level

Start at the **lowest level** of the mark scheme and use it as a **ladder** to see whether the answer meets the descriptor for that level.

When assigning a level you should look at the **overall quality** of the answer as well as the **indicative content**.

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	Level 1: Fragmented ideas, some may be relevant, insufficient links to form an explanation.	1–2		2 AO1/1 6.3.1.1
	No relevant content.	0		



Step 2: Determine a mark

If the response is **predominantly level 2** with a **small amount of level 3** material it would be placed in **level 2** but be awarded a mark at the top of the level because of the level 3 content.

Indicative content

- current in the wire causes heating
- increases temperature of the metal wires / ice

Solid

- arrangement of particles is regular
- particles vibrate about a fixed position

Melting

- internal energy of the ice increases, increasing the temperature to melting point
- so (as the temperature increases) particles vibrate faster
- eventually particles vibrate fast enough to break free from the (strong) bonds
- therefore the arrangement of particles becomes irregular

Liquid

- arrangement of particles is irregular
- particles movement (translational) is random

